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The Covalent Bonding chapter of this Prentice Hall Chemistry Companion Course helps students learn the essential lessons associated with covalent bonding.

### Prentice Hall Chemistry Chapter 8: Covalent Bonding

Chapter 8 Covalent Bonding and Molecular Structure 8-1 Chapter 8: Covalent Bonding and Molecular Structure Chapter In Context In this chapter and the next, we examine chemical bonding in detail.

### Chapter 8: Covalent Bonding and Molecular Structure

Chapter 8 Concepts of Chemical Bonding . Chemical Bonds Three types: Ionic Electrostatic attraction between ions Covalent Sharing of electrons Metallic Metal atoms bonded to several other atoms . Ionic Bonding When a metal and a nonmetal get together . Energetics of Ionic Bonding it takes 495 kJ/mol to remove 1 electron from sodium.  $495 \times 2 = 990 \text{ kJ}$   $2 \text{ Na} + \text{Cl}_2(\text{g}) \rightarrow 2 \text{ NaCl}(\text{s}) \dots$

### Chapter 8 Concepts of Chemical Bonding

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